

SECTION C — CHEMISTRY; METALLURGY

C01 INORGANIC CHEMISTRY

Note(s)

1. In subclasses C01B-C01G, and within each of these subclasses, in the absence of an indication to the contrary, a compound is classified in the last appropriate place, e.g. potassium permanganate is classified only as a permanganate compound, in subclass C01G.
2. Biocidal, pest repellent, pest attractant or plant growth regulatory activity of compounds or preparations is further classified in subclass A01P.

C01B NON-METALLIC ELEMENTS; COMPOUNDS THEREOF (fermentation or enzyme-using processes for the preparation of elements or inorganic compounds except carbon dioxide C12P 3/00; production of non-metallic elements or inorganic compounds by electrolysis or electrophoresis C25B)

Note(s)

1. In this subclass, tradenames that are often found in scientific and patent literature have been used in order to define precisely the scope of the groups.
2. Attention is drawn to the definitions of groups of chemical elements following the title of section C.
3. Attention is drawn to Note (1) after class C01, which defines the last place priority rule applied in this class, i.e. in the range of subclasses C01B-C01G and within these subclasses.
4. Therapeutic activity of compounds is further classified in subclass A61P.

Subclass index

HYDROGEN; HYDROGEN ISOTOPES; WATER; HYDRIDES.....	3/00, 4/00, 5/00, 6/00
SYNTHESIS GAS.....	3/00
HALOGENS OR THEIR COMPOUNDS.....	7/00, 9/00, 11/00
OXYGEN, OXIDES IN GENERAL; PER-COMPOUNDS.....	13/00, 15/00
SULFUR, COMPOUNDS THEREOF.....	17/00
NITROGEN, COMPOUNDS THEREOF.....	21/00
PHOSPHORUS, COMPOUNDS THEREOF.....	25/00
CARBON, COMPOUNDS THEREOF.....	31/00
SILICON, COMPOUNDS THEREOF.....	33/00
SELENIUM OR TELLURIUM; BORON.....	19/00, 35/00
NOBLE GASES.....	23/00
COMPOUNDS HAVING MOLECULAR SIEVE PROPERTIES BUT NOT HAVING BASE-EXCHANGE PROPERTIES.....	37/00
COMPOUNDS HAVING MOLECULAR SIEVE AND BASE-EXCHANGE PROPERTIES.....	39/00

Hydrogen; Hydrides; Water; Synthesis gas from hydrocarbons

	3/12	• • •	by reaction of water vapour with carbon monoxide [3]
3/00 Hydrogen; Gaseous mixtures containing hydrogen; Separation of hydrogen from mixtures containing it; Purification of hydrogen (production of water-gas or synthesis gas from solid carbonaceous material C10J) [3]	3/14	• • • •	Handling of heat and steam [3]
	3/16	• • • •	using catalysts [3]
	3/18	• • • •	using moving solid particles [3]
	3/20	• • • •	by reaction of metal hydroxides with carbon monoxide [3]
3/02 • Production of hydrogen or of gaseous mixtures containing hydrogen [3]	3/22	• •	by decomposition of gaseous or liquid organic compounds [3]
3/04 • • by decomposition of inorganic compounds, e.g. ammonia [3]	3/24	• • •	of hydrocarbons [3]
	3/26	• • •	using catalysts [3]
3/06 • • by reaction of inorganic compounds containing electro-positively bound hydrogen, e.g. water, acids, bases, ammonia, with inorganic reducing agents (by electrolysis of water C25B 1/04) [3]	3/28	• • • •	using moving solid particles [3]
	3/30	• • • •	using the fluidised bed technique [3]
3/08 • • • with metals [3]	3/32	• •	by reaction of gaseous or liquid organic compounds with gasifying agents, e.g. water, carbon dioxide, air [3]
3/10 • • • • by reaction of water vapour with metals [3]			

C01B

- 3/34 • • • by reaction of hydrocarbons with gasifying agents [3]
- 3/36 • • • • using oxygen or mixtures containing oxygen as gasifying agents [3]
- 3/38 • • • • using catalysts [3]
- 3/40 • • • • • characterised by the catalyst [3]
- 3/42 • • • • • using moving solid particles [3]
- 3/44 • • • • • using the fluidised bed technique [3]
- 3/46 • • • • using discontinuously preheated non-moving solid materials, e.g. blast and run [3]
- 3/48 • • • • followed by reaction of water vapour with carbon monoxide [3]
- 3/50 • Separation of hydrogen or hydrogen containing gases from gaseous mixtures, e.g. purification (C01B 3/14 takes precedence) [3]
- 3/52 • • by contacting with liquids; Regeneration of used liquids [3]
- 3/54 • • • including a catalytic reaction [3]
- 3/56 • • by contacting with solids; Regeneration of used solids [3]
- 3/58 • • • including a catalytic reaction [3]

4/00 Hydrogen isotopes; Inorganic compounds thereof prepared by isotope exchange, e.g. $\text{NH}_3 + \text{D}_2 \rightarrow \text{NH}_2\text{D} + \text{HD}$ [2]

5/00 Water

- 5/02 • Heavy water; Preparation by chemical reaction of hydrogen isotopes or their compounds, e.g. $4\text{ND}_3 + 7\text{O}_2 \rightarrow 4\text{NO}_2 + 6\text{D}_2\text{O}$, $2\text{D}_2 + \text{O}_2 \rightarrow 2\text{D}_2\text{O}$

6/00 Hydrides of metals; Monoborane or diborane; Addition complexes thereof [2]

- 6/02 • Hydrides of transition elements; Addition complexes thereof
- 6/04 • Hydrides of alkali metals, alkaline earth metals, beryllium or magnesium; Addition complexes thereof
- 6/06 • Hydrides of aluminium, gallium, indium, thallium, germanium, tin, lead, arsenic, antimony, bismuth or polonium; Monoborane; Diborane; Addition complexes thereof
- 6/10 • • Monoborane; Diborane; Addition complexes thereof [2]
- 6/11 • • • Preparation from boron or inorganic compounds containing boron and oxygen [2]
- 6/13 • • • Addition complexes of monoborane or diborane, e.g. with phosphine, arsine or hydrazine [2]
- 6/15 • • • • Metal borohydrides; Addition complexes thereof [2]
- 6/17 • • • • • Preparation from boron or inorganic compounds containing boron and oxygen [2]
- 6/19 • • • • • Preparation from other compounds of boron [2]
- 6/21 • • • • • Preparation of borohydrides of alkali metals, alkaline earth metals, magnesium or beryllium; Addition complexes thereof, e.g. $\text{LiBH}_4 \cdot 2\text{N}_2\text{H}_4$, NaB_2H_7 [2]
- 6/23 • • • • • Preparation of borohydrides of other metals, e.g. aluminium borohydride; Addition complexes thereof, e.g. $\text{Li}[\text{Al}(\text{BH}_4)_3\text{H}]$ [2]
- 6/24 • Hydrides containing at least two metals, e.g. $\text{Li}(\text{AlH}_4)$; Addition complexes thereof (C01B 6/13-C01B 6/23 take precedence) [2]

- 6/26 • • Preparation from the metal with the highest valency or from its oxides or salts of its oxyacids
- 6/34 • Purification; Stabilisation

Halogens; Compounds thereof

7/00 Halogens; Halogen acids

- 7/01 • Chlorine; Hydrogen chloride [2]
- 7/03 • • Preparation from chlorides [2, 3]
- 7/04 • • • Preparation of chlorine from hydrogen chloride [3]
- 7/05 • • • Preparation from ammonium chloride [2, 3]
- 7/07 • • Purification [2, 3]
- 7/075 • • • of liquid chlorine [2, 3]
- 7/09 • Bromine; Hydrogen bromide [2]
- 7/13 • Iodine; Hydrogen iodide [2]
- 7/14 • • Iodine [2]
- 7/16 • • • Preparation from seaweed [2]
- 7/19 • Fluorine; Hydrogen fluoride [2]
- 7/20 • • Fluorine [2]
- 7/24 • Inter-halogen compounds

9/00 General methods of preparing halides (particular individual halides, see the relevant groups in subclasses C01B-C01G according to the element combined with the halogen; electrolytic production of inorganic compounds C25B)

- 9/02 • Chlorides
- 9/04 • Bromides
- 9/06 • Iodides
- 9/08 • Fluorides

11/00 Oxides or oxyacids of halogens; Salts thereof

- 11/02 • Oxides of chlorine
- 11/04 • Hypochlorous acid
- 11/06 • • Hypochlorites, e.g. chlorinated lime
- 11/08 • Chlorous acid
- 11/10 • • Chlorites
- 11/12 • Chloric acid
- 11/14 • • Chlorates
- 11/16 • Perchloric acid
- 11/18 • • Perchlorates
- 11/20 • Oxygen compounds of bromine
- 11/22 • Oxygen compounds of iodine
- 11/24 • Oxygen compounds of fluorine

Oxygen; Oxides or hydroxides in general; Per-compounds

13/00 Oxygen; Ozone; Oxides or hydroxides in general

- 13/02 • Preparation of oxygen (by liquefying F25J)
- 13/08 • • from air with the aid of metal oxides, e.g. barium oxide, manganese oxide
- 13/10 • Preparation of ozone
- 13/11 • • by electric discharge [2]
- 13/14 • Methods for preparing oxides or hydroxides in general (particular individual oxides or hydroxides, see the relevant groups of subclasses C01B-C01G or C25B, according to the element combined with the oxygen or hydroxy group)
- 13/16 • • Purification [3]
- 13/18 • • by thermal decomposition of compounds, e.g. of salts or hydroxides [3]

- 13/20 • • by oxidation of elements in the gaseous state; by oxidation or hydrolysis of compounds in the gaseous state [3]
- 13/22 • • • of halides or oxyhalides [3]
- 13/24 • • • • in the presence of hot combustion gases [3]
- 13/26 • • • • in the presence of a fluidised bed [3]
- 13/28 • • • • using a plasma or an electric discharge [3]
- 13/30 • • • • Removal and cooling of the oxide containing suspension [3]
- 13/32 • • by oxidation or hydrolysis of elements or compounds in the liquid or solid state [3]
- 13/34 • • by oxidation or hydrolysis of sprayed or atomised solutions [3]
- 13/36 • • by precipitation reactions in solutions [3]
- 15/00 Peroxides; Peroxyhydrates; Peroxyacids or salts thereof; Superoxides; Ozonides**
- 15/01 • Hydrogen peroxide [3]
- 15/013 • • Separation; Purification; Concentration [3]
- 15/017 • • • Anhydrous hydrogen peroxide; Anhydrous solutions or gaseous mixtures containing hydrogen peroxide [3]
- 15/022 • • Preparation from organic compounds [2]
- 15/023 • • • by the alkyl-anthraquinone process [3]
- 15/024 • • • from hydrocarbons [3]
- 15/026 • • • from alcohols [3]
- 15/027 • • Preparation from water [3]
- 15/029 • • Preparation from hydrogen and oxygen [3]
- 15/03 • • Preparation from inorganic peroxy-compounds, e.g. from peroxysulfates [3]
- 15/032 • • • from metal peroxides [3]
- 15/037 • • Stabilisation by additives [3]
- 15/04 • Metal peroxides or peroxyhydrates thereof; Superoxides; Ozonides [3]
- 15/043 • • of alkali metals, alkaline earth metals or of magnesium [2, 3]
- 15/047 • • of heavy metals [2, 3]
- 15/055 • Peroxyhydrates (C01B 15/04 takes precedence); Peroxyacids or salts thereof [3]
- 15/06 • • containing sulfur [3]
- 15/08 • • • Peroxysulfates [3]
- 15/10 • • containing carbon [3]
- 15/12 • • containing boron [3]
- 15/14 • • containing silicon [3]
- 15/16 • • containing phosphorus [3]
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- 17/00 Sulfur; Compounds thereof**
- 17/02 • Preparation of sulfur; Purification
- 17/027 • • Recovery of sulfur from material containing elemental sulfur, e.g. luxmasses; Purification [3]
- 17/033 • • • using a liquid extractant [3]
- 17/04 • • from gaseous sulfur compounds including gaseous sulfides
- 17/05 • • • by wet processes [3]
- 17/06 • • from non-gaseous sulfides or materials containing such sulfides, e.g. ores
- 17/10 • • Finely-divided sulfur, e.g. sublimed sulfur, flowers of sulfur
- 17/12 • • Insoluble sulfur (μ -sulfur)
- 17/16 • Hydrogen sulfides
- 17/18 • • Hydrogen polysulfides
- 17/20 • Methods for preparing sulfides or polysulfides, in general (ammonium sulfides or polysulfides C01C; sulfides or polysulfides of metals, other than alkali metals, magnesium, calcium, strontium, and barium, see the relevant groups of subclasses C01F or C01G, according to the metal)
- 17/22 • Alkali metal sulfides or polysulfides
- 17/24 • • Preparation by reduction
- 17/26 • • • with carbon
- 17/28 • • • with reducing gases
- 17/30 • • Preparation from sodium or potassium amalgam with sulfur or sulfides
- 17/32 • • Hydrosulfides of sodium or potassium
- 17/34 • • Polysulfides of sodium or potassium
- 17/36 • • Purification
- 17/38 • • Dehydration
- 17/40 • • Making shaped products, e.g. granules
- 17/42 • Sulfides or polysulfides of magnesium, calcium, strontium, or barium
- 17/43 • • from oxides or hydroxides with sulfur or hydrogen sulfide
- 17/44 • • by reduction of sulfates
- 17/45 • Compounds containing sulfur and halogen, with or without oxygen
- 17/46 • Compounds containing sulfur, halogen, hydrogen, and oxygen
- 17/48 • Sulfur dioxide; Sulfurous acid
- 17/50 • • Preparation of sulfur dioxide
- 17/52 • • • by roasting sulfides (C22B 1/00 takes precedence)
- 17/54 • • • by burning elemental sulfur
- 17/56 • • • Separation; Purification
- 17/58 • • • Recovery of sulfur dioxide from acid tar or the like
- 17/60 • • • Isolation of sulfur dioxide from gases
- 17/62 • Methods of preparing sulfites in general (particular individual sulfites, see the relevant groups of subclasses C01B-C01G, according to the cation)
- 17/64 • Thiosulfates; Dithionites; Polythionates
- 17/66 • • Dithionites
- 17/69 • Sulfur trioxide; Sulfuric acid [3]
- 17/70 • • Stabilisation of gamma-form sulfur trioxide
- 17/74 • • Preparation [3]
- 17/76 • • • by contact processes
- 17/765 • • • • Multi-stage SO_3 -conversion [3]
- 17/77 • • • • Fluidised-bed processes [3]
- 17/775 • • • • Liquid phase contacting processes or wet catalysis processes [3]
- 17/78 • • • • characterised by the catalyst used
- 17/79 • • • • containing vanadium [3]
- 17/80 • • • • Apparatus
- 17/82 • • • of sulfuric acid using a nitrogen oxide process
- 17/84 • • • • Chamber process
- 17/86 • • • • Tower process
- 17/88 • • Concentration of sulfuric acid
- 17/90 • • Separation; Purification
- 17/92 • • • Recovery from acid tar or the like
- 17/94 • • • Recovery from nitration acids
- 17/96 • Methods for the preparation of sulfates in general (particular individual sulfates, see the relevant groups of subclasses C01B-C01G, according to the cation)
- 17/98 • Other compounds containing sulfur and oxygen (persulfuric acids C01B 15/06; persulfates C01B 15/08)

C01B

- 19/00 Selenium; Tellurium; Compounds thereof**
 19/02 • Elemental selenium or tellurium [3]
 19/04 • Binary compounds [3]
- 21/00 Nitrogen; Compounds thereof**
 21/02 • Preparation of nitrogen (by decomposition of ammonia C01B 3/04)
 21/04 • Purification or separation of nitrogen (by liquefying F25J)
 21/06 • Binary compounds of nitrogen with metals, with silicon, or with boron
 21/064 • • with boron [3]
 21/068 • • with silicon [3]
 21/072 • • with aluminium [3]
 21/076 • • with titanium or zirconium [3]
 21/08 • Hydrazoic acid; Azides; Halogen azides
 21/082 • Compounds containing nitrogen and non-metals (C01B 21/06, C01B 21/08 take precedence) [3]
 21/083 • • containing one or more halogen atoms [3]
 21/084 • • • containing also one or more oxygen atoms, e.g. nitrosyl halides [3]
 21/086 • • containing one or more sulfur atoms [3]
 21/087 • • containing one or more hydrogen atoms [3]
 21/088 • • • containing also one or more halogen atoms [3]
 21/09 • • • Halogeno-amines, e.g. chloramine [3]
 21/092 • • • containing also one or more metal atoms [3]
 21/093 • • • containing also one or more sulfur atoms [3]
 21/094 • • • Nitrosyl containing acids [3]
 21/096 • • • Amidosulfonic acid; Salts thereof [3]
 21/097 • • containing phosphorus atoms [3]
 21/098 • • • Phosphonitrilic dihalides; Polymers thereof [3]
 21/12 • Carbamic acid; Salts thereof
 21/14 • Hydroxylamine; Salts thereof
 21/16 • Hydrazine; Salts thereof
 21/20 • Nitrogen oxides; Oxyacids of nitrogen; Salts thereof
 21/22 • Nitrous oxide (N₂O)
 21/24 • Nitric oxide (NO)
 21/26 • • Preparation by catalytic oxidation of ammonia
 21/28 • • • Apparatus
 21/30 • • Preparation by oxidation of nitrogen
 21/32 • • • Apparatus
 21/34 • • Nitrogen trioxide (N₂O₃)
 21/36 • • Nitrogen dioxide (NO₂, N₂O₄) (C01B 21/26, C01B 21/30 take precedence)
 21/38 • • Nitric acid
 21/40 • • • Preparation by absorption of oxides of nitrogen
 21/42 • • • Preparation from nitrates
 21/44 • • • Concentration
 21/46 • • • Purification; Separation
 21/48 • • Methods for the preparation of nitrates in general (particular individual nitrates, see the relevant groups of subclasses C01B-C01G, according to the cation)
 21/50 • • Nitrous acid; Salts thereof
- 23/00 Noble gases; Compounds thereof (liquefying F25J)**
- 25/00 Phosphorus; Compounds thereof (C01B 21/00, C01B 23/00 take precedence; perphosphates C01B 15/16) [3]**
 25/01 • Treating phosphate ores or other raw phosphate materials to obtain phosphorus or phosphorus compounds [2]
 25/02 • Preparation of phosphorus
 25/023 • • of red phosphorus [2]
 25/027 • • of yellow phosphorus [2]
 25/04 • Purification of phosphorus
 25/043 • • of red phosphorus [2]
 25/047 • • of yellow phosphorus [2]
 25/06 • Hydrogen phosphides
 25/08 • Other phosphides
 25/10 • Halides or oxyhalides of phosphorus [2]
 25/12 • Oxides of phosphorus
 25/14 • Sulfur, selenium, or tellurium compounds of phosphorus
 25/16 • Oxyacids of phosphorus; Salts thereof (peroxyacids or salts thereof C01B 15/00)
 25/163 • • Phosphorous acid; Salts thereof [2]
 25/165 • • Hypophosphorous acid; Salts thereof [2]
 25/168 • • Pyrophosphorous acid; Salts thereof [2]
 25/18 • • Phosphoric acid
 25/20 • • • Preparation from elemental phosphorus or phosphoric anhydride
 25/22 • • • Preparation by reacting phosphate containing material with an acid, e.g. wet process
 25/222 • • • • with sulfuric acid, a mixture of acids mainly consisting of sulfuric acid or a mixture of compounds forming it *in situ*, e.g. a mixture of sulfur dioxide, water and oxygen [3]
 25/223 • • • • only one form of calcium sulfate being formed [3]
 25/225 • • • • Dihydrate process [3]
 25/226 • • • • Hemihydrate process [3]
 25/228 • • • • one form of calcium sulfate being formed and then converted to another form [3]
 25/229 • • • • Hemihydrate-dihydrate process [3]
 25/231 • • • • Dihydrate-hemihydrate process [3]
 25/232 • • • • Preparation by reacting phosphate containing material with concentrated sulfuric acid and subsequently lixiviating the obtained mass, e.g. clinker process [3]
 25/234 • • • Purification; Stabilisation; Concentration (purification concomitant with preparation C01B 25/22; preparation involving solvent-solvent extraction C01B 25/46) [3]
 25/235 • • • Clarification; Stabilisation to prevent post-precipitation of dissolved impurities [3]
 25/237 • • • Selective elimination of impurities [3]
 25/238 • • • Cationic impurities [3]
 25/24 • • Condensed phosphoric acids
 25/26 • • Phosphates (perphosphates C01B 15/16)
 25/28 • • Ammonium phosphates
 25/30 • • Alkali metal phosphates
 25/32 • • Phosphates of magnesium, calcium, strontium, or barium
 25/34 • • • Magnesium phosphates
 25/36 • • • Aluminium phosphates
 25/37 • • • Phosphates of heavy metals [2]
 25/38 • • • Condensed phosphates
 25/39 • • • • of alkali metals [3]
 25/40 • • • • Polyphosphates [2]
 25/41 • • • • of alkali metals [3]
 25/42 • • • • Pyrophosphates [2]
 25/44 • • • • Metaphosphates [2]
 25/445 • • • • of alkali metals [3]
 25/45 • • • containing plural metal, or metal and ammonium [3]
 25/455 • • • containing halogen [3]

- 25/46 • • Preparation involving solvent-solvent extraction [2]
- 31/00 Carbon; Compounds thereof** (C01B 21/00, C01B 23/00 take precedence; percarbonates C01B 15/10; carbon black C09C 1/48) [3]
- 31/02 • Preparation of carbon (by using ultra-high pressure, e.g. for the formation of diamonds, B01J 3/06; by crystal growth C30B); Purification
- 31/04 • • Graphite
- 31/06 • • Diamond
- 31/08 • Active carbon
- 31/10 • • Preparation by using gaseous activating agents
- 31/12 • • Preparation by using non-gaseous activating agents
- 31/14 • • Granulation
- 31/16 • Preparation of ion-exchanging materials from carbonaceous material
- 31/18 • Carbon monoxide
- 31/20 • Carbon dioxide
- 31/22 • • Solidifying
- 31/24 • Methods for the preparation of carbonates or bicarbonates in general (percarbonates C01B 15/10; particular individual carbonates, see the relevant groups of subclasses C01B-C01G, according to the cation)
- 31/26 • Compounds containing carbon and sulfur, e.g. carbon disulfide, carbon oxysulfide; Thiophosgene
- 31/28 • Phosgene
- 31/30 • Carbides
- 31/32 • • Calcium carbide
- 31/34 • • Tungsten or molybdenum carbides
- 31/36 • • Carbides of silicon or boron
- 33/00 Silicon; Compounds thereof** (C01B 21/00, C01B 23/00 take precedence; persilicates C01B 15/14; carbides C01B 31/36) [3]
- 33/02 • Silicon (forming single crystals or homogeneous polycrystalline material with defined structure C30B) [5]
- 33/021 • • Preparation (chemical coating from the vapour phase C23C 16/00) [5]
- 33/023 • • • by reduction of silica or silica-containing material [5]
- 33/025 • • • • with carbon or a solid carbonaceous material, i.e. carbo-thermal process [5]
- 33/027 • • • by decomposition or reduction of gaseous or vaporised silicon compounds other than silica or silica-containing material [5]
- 33/029 • • • • by decomposition of monosilane [5]
- 33/03 • • • • by decomposition of silicon halides or halosilanes or reduction thereof with hydrogen as the only reducing agent [5]
- 33/031 • • • • • by decomposition of silicon tetraiodide [5]
- 33/033 • • • • by reduction of silicon halides or halosilanes with a metal or a metallic alloy as the only reducing agents [5]
- 33/035 • • • • by decomposition or reduction of gaseous or vaporised silicon compounds in the presence of heated filaments of silicon, carbon or a refractory metal, e.g. tantalum or tungsten, or in the presence of heated silicon rods on which the formed silicon is deposited, a silicon rod being obtained, e.g. Siemens process [5]
- 33/037 • • Purification (by zone-melting C30B 13/00) [5]
- 33/039 • • • by conversion of the silicon into a compound, optional purification of the compound, and reconversion into silicon [5]
- 33/04 • Hydrides of silicon
- 33/06 • Metal silicides
- 33/08 • Compounds containing halogen
- 33/10 • • Compounds containing silicon, fluorine, and other elements
- 33/107 • • Halogenated silanes [3]
- 33/113 • Silicon oxides; Hydrates thereof [3]
- 33/12 • • Silica; Hydrates thereof, e.g. lepidic silicic acid [3]
- 33/14 • • • Colloidal silica, e.g. dispersions, gels, sols [3]
- 33/141 • • • • Preparation of hydrosols or aqueous dispersions [3]
- 33/142 • • • • • by acidic treatment of silicates [3]
- 33/143 • • • • • of aqueous solutions of silicates [3]
- 33/145 • • • • Preparation of hydroorganosols, organosols or dispersions in an organic medium [3]
- 33/146 • • • • After-treatment of sols (preparation of hydroorganosols, organosols or dispersions in an organic medium from hydrosols C01B 33/145) [3]
- 33/148 • • • • • Concentration; Drying; Dehydration; Stabilisation; Purification [3]
- 33/149 • • • • • Coating [3]
- 33/151 • • • • • by progressively adding a sol to a different sol, i.e. "build up" of particles using a "heel" [3]
- 33/152 • • • • • Preparation of hydrogels [3]
- 33/154 • • • • • by acidic treatment of aqueous silicate solutions [3]
- 33/155 • • • • • Preparation of hydroorganogels or organogels [3]
- 33/157 • • • • • After-treatment of gels [3]
- 33/158 • • • • • Purification; Drying; Dehydrating [3]
- 33/159 • • • • • Coating or hydrophobisation [3]
- 33/16 • • • Preparation of silica xerogels [3]
- 33/18 • • • Preparation of finely divided silica neither in sol nor in gel form; After-treatment thereof (treatment to enhance the pigmenting or filling properties C09C) [3]
- 33/187 • • • • • by acidic treatment of silicates [3]
- 33/193 • • • • • of aqueous solutions of silicates [3]
- 33/20 • Silicates (persilicates C01B 15/14)
- 33/22 • • Magnesium silicates
- 33/24 • • Alkaline earth metal silicates
- 33/26 • • Aluminium-containing silicates [5]
- 33/32 • • Alkali metal silicates (C01B 33/26 takes precedence) [3]
- 33/36 • • having base-exchange properties but not having molecular sieve properties [6]
- 33/38 • • • Layered base-exchange silicates, e.g. clays, micas or alkali metal silicates of kenyaite or magadiite type [6]
- 33/40 • • • • Clays [6]
- 33/42 • • • • Micas [6]
- 33/44 • • • • Products obtained from layered base-exchange silicates by ion-exchange with organic compounds such as ammonium, phosphonium or sulfonium compounds or by intercalation of organic compounds, e.g. organoclay material [6]
- 33/46 • • • Amorphous silicates, e.g. so-called "amorphous zeolites" [6]

- 35/00 Boron; Compounds thereof** (monoborane, diborane, metal borohydrides or addition complexes thereof C01B 6/00; perborates C01B 15/12; binary compounds with nitrogen C01B 21/06; phosphides C01B 25/08; carbides C01B 31/36) [2]
- 35/02 • Boron; Borides [2]
- 35/04 • • Metal borides [2]
- 35/06 • Boron halogen compounds [2]
- 35/08 • Compounds containing boron and nitrogen, phosphorus, oxygen, sulfur, selenium or tellurium [2]
- 35/10 • • Compounds containing boron and oxygen (C01B 35/06 takes precedence) [2]
- 35/12 • • • Borates [2]
- 35/14 • • Compounds containing boron and nitrogen, phosphorus, sulfur, selenium or tellurium [2]
- 35/16 • Compounds containing direct bonding between two boron atoms, e.g. $\text{Cl}_2\text{B}-\text{BCl}_2$ [2]
- 35/18 • Compounds containing three or more boron atoms, e.g. NaB_3H_8 , $\text{MgB}_{10}\text{Br}_{10}$ (borazoles C01B 35/14) [2]

Compounds characterised primarily by their physical or chemical properties, rather than by their chemical constitution [6]

- 37/00 Compounds having molecular sieve properties but not having base-exchange properties [6]**
- 37/02 • Crystalline silica-polymorphs, e.g. silicalites [6]
- 37/04 • Aluminophosphates (APO compounds) [6]
- 37/06 • Aluminophosphates containing other elements, e.g. metals, boron [6]
- 37/08 • • Silicoaluminophosphates (SAPO compounds) [6]
- 39/00 Compounds having molecular sieve and base-exchange properties, e.g. crystalline zeolites; Their preparation; After-treatment, e.g. ion-exchange or dealumination** (treatment to modify the sorption properties, e.g. shaping using a binder, B01J 20/10; treatment to modify the catalytic properties, e.g. combination of treatments to make the zeolites appropriate to their use as a catalyst, B01J 29/04; treatment to improve the ion-exchange properties B01J 39/14) [6]

Note(s)

In this group, the following term is used with the meaning indicated:

- "zeolites" means:
 - i. crystalline aluminosilicates with base-exchange and molecular sieve properties, having three dimensional, microporous lattice framework structure of tetrahedral oxide units;
 - ii. compounds isomorphous to those of the former category, wherein the aluminium or silicon atoms in the framework are partly or wholly replaced by atoms of other elements, e.g. by gallium, germanium, phosphorus or boron.

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- 39/02 • Crystalline aluminosilicate zeolites; Isomorphous compounds thereof; Direct preparation thereof; Preparation thereof starting from a reaction mixture containing a crystalline zeolite of another type, or from preformed reactants: After-treatment thereof [6]
- 39/04 • • using at least one organic template directing agent, e.g. an ionic quaternary ammonium compound or an aminated compound [6]
- 39/06 • • Preparation of isomorphous zeolites characterised by measures to replace the aluminium or silicon atoms in the lattice framework by atoms of other elements [6]
- 39/08 • • • the aluminium atoms being wholly replaced [6]
- 39/10 • • • the replacing atoms being phosphorus atoms [6]
- 39/12 • • • the replacing atoms being boron atoms [6]
- 39/14 • • Type A [6]
- 39/16 • • • from aqueous solutions of an alkali metal aluminate and an alkali metal silicate excluding any other source of alumina or silica but seeds [6]
- 39/18 • • • from a reaction mixture containing at least one aluminium silicate or aluminosilicate of a clay type, e.g. kaolin or metakaolin or its exotherm modification or allophane [6]
- 39/20 • • Faujasite type, e.g. type X or Y [6]
- 39/22 • • • Type X [6]
- 39/24 • • • Type Y [6]
- 39/26 • • Mordenite type [6]
- 39/28 • • Phillipsite or harmotome type, e.g. type B [6]
- 39/30 • • Erionite or offretite type, e.g. zeolite T [6]
- 39/32 • • Type L [6]
- 39/34 • • Type ZSM-4 or type Ω [6]
- 39/36 • • Pentasil type, e.g. types ZSM-5, ZSM-8 or ZSM-11 [6]
- 39/38 • • • Type ZSM-5 [6]
- 39/40 • • • using at least one organic template directing agent [6]
- 39/42 • • Type ZSM-12 [6]
- 39/44 • • Ferrierite type, e.g. types ZSM-21, ZSM-35 or ZSM-38 [6]
- 39/46 • • Other types characterised by their X-ray diffraction pattern and their defined composition [6]
- 39/48 • • • using at least one organic template directing agent [6]
- 39/50 • Zeolites wherein inorganic bases or salts occlude channels in the lattice framework, e.g. sodalite, cancrinite, nosean, hauynite [6]
- 39/52 • • Sodalites [6]
- 39/54 • Phosphates, e.g. APO or SAPO compounds [6]

C01C AMMONIA; CYANOGEN; COMPOUNDS THEREOF (salts of oxyacids of halogens C01B 11/00; peroxides, salts of peroxyacids C01B 15/00; thiosulfates, dithionites, polythionates C01B 17/64; compounds containing selenium or tellurium C01B 19/00; azides C01B 21/08; metal amides C01B 21/092; nitrites C01B 21/50; phosphides C01B 25/08; salts of oxyacids of phosphorus C01B 25/16; compounds containing silicon C01B 33/00; compounds containing boron C01B 35/00; fermentation or enzyme-using processes for the preparation of elements or inorganic compounds except carbon dioxide C12P 3/00; production of non-metallic elements or inorganic compounds by electrolysis or electrophoresis C25B)

Note(s)

- Attention is drawn to Note (1) after class C01, which defines the last place priority rule applied in this class, i.e. in the range of subclasses C01B-C01G and within these subclasses.
- Therapeutic activity of compounds is further classified in subclass A61P.

1/00 Ammonia; Compounds thereof	1/26	• Carbonates or bicarbonates of ammonium
1/02 • Preparation or separation of ammonia	1/28	• Methods of preparing ammonium salts in general
1/04 • • Preparation of ammonia by synthesis (preparation or purification of gas mixtures for ammonia synthesis C01B 3/02)		
1/08 • • Preparation of ammonia from nitrogenous organic substances		
1/10 • • Separation of ammonia from ammonia liquors, e.g. gas liquors		
1/12 • • Separation of ammonia from gases and vapours		
1/14 • • • Saturators		
1/16 • Halides of ammonium		
1/18 • Nitrates of ammonium		
1/20 • Sulfides; Polysulfides		
1/22 • Sulfites of ammonium		
1/24 • Sulfates of ammonium (C01C 1/14 takes precedence)	3/00 Cyanogen; Compounds thereof	
1/242 • • Preparation from ammonia and sulfuric acid or sulfur trioxide [2]	3/02	• Preparation of hydrogen cyanide
1/244 • • Preparation by double decomposition of ammonium salts with sulfates [2]	3/04	• • Separation from gases
1/245 • • Preparation from compounds containing nitrogen and sulfur [2]	3/06	• Stabilisation of hydrogen cyanide
1/246 • • • from sulfur-containing ammonium compounds [2]	3/08	• Simple or complex cyanides of metals
1/247 • • • • by oxidation with free oxygen [2]	3/10	• • Simple alkali metal cyanides [3]
1/248 • • Preventing coalescing or controlling form or size of crystals [2]	3/11	• • Complex cyanides [3]
1/249 • • Deacidifying the crystals [2]	3/12	• • Simple or complex iron cyanides [2]
	3/14	• Cyanic acid; Salts thereof
	3/16	• Cyanamide; Salts thereof
	3/18	• • Calcium cyanamide
	3/20	• Thiocyanic acid; Salts thereof

Note(s)

- This group does not cover ammonium salts of complex acids (other than complex cyanides) containing a metal in the anion, which are covered by the relevant groups of subclasses C01D-C01G, according to the metal.
- Salts of polybasic acids with ammonium and a metal as cations are classified as though the ammonium were hydrogen.
- Complex ammine salts are classified in the relevant groups of subclasses C01D-C01G, according to the metal.

C01D COMPOUNDS OF ALKALI METALS, i.e. LITHIUM, SODIUM, POTASSIUM, RUBIDIUM, CAESIUM, OR FRANCIUM (metal hydrides C01B 6/00; salts of oxyacids of halogens C01B 11/00; peroxides, salts of peroxyacids C01B 15/00; sulfides or polysulfides C01B 17/22; thiosulfates, dithionites, polythionates C01B 17/64; compounds containing selenium or tellurium C01B 19/00; binary compounds of nitrogen with metals C01B 21/06; azides C01B 21/08; metal amides C01B 21/092; nitrites C01B 21/50; phosphides C01B 25/08; salts of oxyacids of phosphorus C01B 25/16; carbides C01B 31/30; compounds containing silicon C01B 33/00; compounds containing boron C01B 35/00; cyanides C01C 3/08; salts of cyanic acid C01C 3/14; salts of cyanamide C01C 3/16; thiocyanates C01C 3/20; fermentation or enzyme-using processes for the preparation of elements or inorganic compounds except carbon dioxide C12P 3/00; obtaining metal compounds from mixtures, e.g. ores, which are intermediate compounds in a metallurgical process for obtaining a free metal C22B; production of non-metallic elements or inorganic compounds by electrolysis or electrophoresis C25B)

Note(s)

- Attention is drawn to Note (1) after class C01, which defines the last place priority rule applied in this class, i.e. in the range of subclasses C01B-C01G and within these subclasses.
- Therapeutic activity of compounds is further classified in subclass A61P.

1/00 Oxides or hydroxides of sodium, potassium, or alkali metals in general [2]	1/20	• • Preparation by reacting oxides or hydroxides with alkali metal salts
1/02 • Oxides	1/22	• • • with carbonates or bicarbonates
1/04 • Hydroxides	1/24	• • • from or <u>via</u> fluorides or silico-fluorides

C01D

- 1/26 • • Preparation from or via cyano compounds, e.g. cyanides, cyanamides
- 1/28 • • Purification; Separation
- 1/30 • • • by crystallisation
- 1/32 • • • by adsorption or precipitation
- 1/34 • • • with selective solvents
- 1/36 • • • by oxidation
- 1/38 • • • by dialysis
- 1/40 • • • by electrolysis
- 1/42 • • Concentration; Dehydration
- 1/44 • • Preparation in the form of granules, pieces, or other shaped products
- 3/00 Halides of sodium, potassium, or alkali metals in general [2]**
- 3/02 • Fluorides
- 3/04 • Chlorides
- 3/06 • • Preparation by working up brines, seawater or spent lyes
- 3/08 • • Preparation by working up natural or industrial salt mixtures or siliceous minerals
- 3/10 • Bromides
- 3/12 • Iodides
- 3/14 • Purification
- 3/16 • • by precipitation or adsorption
- 3/18 • • with selective solvents
- 3/20 • • by melting
- 3/22 • Preparation in the form of granules, pieces, or other shaped products
- 3/24 • • Influencing the crystallisation process
- 3/26 • Preventing the absorption of moisture or caking of the crystals
- 5/00 Sulfates or sulfites of sodium, potassium, or alkali metals in general [2]**
- 5/02 • Preparation of sulfates from alkali metal salts and sulfuric acid or bisulfates; Preparation of bisulfates
- 5/04 • Preparation of sulfates with the aid of sulfurous acid or sulfites, e.g. Hargreaves process
- 5/06 • Preparation of sulfates by double decomposition
- 5/08 • • with each other or with ammonium sulfate
- 5/10 • • with sulfates of magnesium, calcium, strontium, or barium
- 5/12 • Preparation of double sulfates of magnesium with sodium or potassium [2]
- 5/14 • Preparation of sulfites (C01D 5/04 takes precedence)
- 5/16 • Purification
- 5/18 • Dehydration
- 7/00 Carbonates of sodium, potassium, or alkali metals in general [2]**
- 7/02 • Preparation by double decomposition
- 7/04 • • with a fluoride or silico-fluoride (C01D 1/24 takes precedence)
- 7/06 • Preparation via sodium or potassium magnesium carbonate
- 7/07 • Preparation from the hydroxides [2]
- 7/08 • Preparation from or via cyano compounds of sodium or potassium (C01D 1/26 takes precedence)
- 7/10 • Preparation of bicarbonates from carbonates (ammonia-soda process C01D 7/18)
- 7/12 • Preparation of carbonates from bicarbonates
- 7/14 • Preparation of sesquicarbonates
- 7/16 • Preparation from compounds of sodium or potassium with amines and carbon dioxide
- 7/18 • Preparation by the ammonia-soda process
- 7/22 • Purification
- 7/24 • • Crystallisation
- 7/26 • • by precipitation or adsorption
- 7/28 • • with selective solvents
- 7/30 • • by oxidation
- 7/32 • • by dialysis
- 7/34 • • by electrolysis
- 7/35 • Varying the content of water of crystallisation or the specific gravity [2]
- 7/37 • • Densifying sodium carbonate [2]
- 7/38 • Preparation in the form of granules, pieces, or other shaped products
- 7/40 • • Influencing the crystallisation process
- 7/42 • Preventing the absorption of moisture or caking
- 9/00 Nitrates of sodium, potassium, or alkali metals in general [2]**
- 9/02 • Preparation by working-up natural salt mixtures
- 9/04 • Preparation with liquid nitric acid
- 9/06 • Preparation with gaseous nitric acid or nitrogen oxides
- 9/08 • Preparation by double decomposition
- 9/10 • • with ammonium nitrate
- 9/12 • • with nitrates of magnesium, calcium, strontium, or barium
- 9/14 • • of salts of potassium with sodium nitrate
- 9/16 • Purification
- 9/18 • Preparation in the form of shaped products, e.g. granules
- 9/20 • Preventing the absorption of moisture or caking
- 13/00 Compounds of sodium or potassium not provided for elsewhere [2]**
- 15/00 Lithium compounds [2]**
- 15/02 • Oxides; Hydroxides [2]
- 15/04 • Halides [2]
- 15/06 • Sulfates; Sulfites [2]
- 15/08 • Carbonates; Bicarbonates [2]
- 15/10 • Nitrates [2]
- 17/00 Rubidium, caesium, or francium compounds [2]**

C01F COMPOUNDS OF THE METALS BERYLLIUM, MAGNESIUM, ALUMINIUM, CALCIUM, STRONTIUM, BARIUM, RADIUM, THORIUM, OR OF THE RARE-EARTH METALS (metal hydrides C01B 6/00; salts of oxyacids of halogens C01B 11/00; peroxides, salts of peroxyacids C01B 15/00; sulfides or polysulfides of magnesium, calcium, strontium, or barium C01B 17/42; thiosulfates, dithionites, polythionates C01B 17/64; compounds containing selenium or tellurium C01B 19/00; binary compounds of nitrogen with metals C01B 21/06; azides C01B 21/08; metal amides C01B 21/092; nitrites C01B 21/50; phosphides C01B 25/08; salts of oxyacids of phosphorus C01B 25/16; carbides C01B 31/30; compounds containing silicon C01B 33/00; compounds containing boron C01B 35/00; compounds having molecular sieve properties but not having base-exchange properties C01B 37/00; compounds having molecular sieve and base-exchange properties, e.g. crystalline zeolites, C01B 39/00; cyanides C01C 3/08; salts of cyanic acid C01C 3/14; salts of cyanamide C01C 3/16; thiocyanates C01C 3/20; fermentation or enzyme-using processes for the preparation of elements or inorganic compounds except carbon dioxide C12P 3/00; obtaining metal compounds from mixtures, e.g. ores, which are intermediate compounds in a metallurgical process for obtaining a free metal C22B; production of non-metallic elements or inorganic compounds by electrolysis or electrophoresis C25B)

Note(s)

- Attention is drawn to Note (1) after class C01, which defines the last place priority rule applied in this class, i.e. in the range of subclasses C01B-C01G and within these subclasses.
- Therapeutic activity of compounds is further classified in subclass A61P.

1/00	Methods of preparing compounds of the metals beryllium, magnesium, aluminium, calcium, strontium, barium, radium, thorium, or the rare earths, in general	7/06	• • • by treating aluminous minerals with alkali hydroxide
		7/08	• • • by treating aluminous minerals with sodium carbonate
3/00	Compounds of beryllium	7/10	• • • by treating aluminous minerals with alkali sulfates and reducing agents
3/02	• Oxides; Hydroxides [3]	7/12	• • • Alkali metal aluminates from alkaline earth metal aluminates
5/00	Compounds of magnesium	7/14	• • • Aluminium oxide or hydroxide from alkali metal aluminates
5/02	• Magnesia	7/16	• • Preparation of alkaline earth metal aluminates; Aluminium oxide or hydroxide therefrom
5/04	• • by oxidation of metallic magnesium	7/18	• • • Aluminium oxide or hydroxide from alkaline earth metal aluminates
5/06	• • by thermal decomposition of magnesium compounds (calcining magnesite or dolomite C04B 2/10)	7/20	• • Preparation of aluminium oxide or hydroxide from aluminous ores with acids or salts
5/08	• • • by calcining magnesium hydroxide	7/22	• • • with halides
5/10	• • • by thermal decomposition of magnesium chloride with water vapour	7/24	• • • with nitric acid or nitrogen oxides
5/12	• • • by thermal decomposition of magnesium sulfate, with or without reduction	7/26	• • • with sulfuric acids or sulfates
5/14	• Magnesium hydroxide	7/28	• • • with sulfurous acid
5/16	• • by treating magnesia, e.g. calcined dolomite, with water or solutions of salts not containing magnesium	7/30	• • Preparation of aluminium oxide or hydroxide by thermal decomposition of aluminium compounds
5/20	• • by precipitation from solutions of magnesium salts with ammonia	7/32	• • • of sulfates
5/22	• • from magnesium compounds with alkali hydroxides or alkaline earth oxides or hydroxides	7/34	• • Preparation of aluminium hydroxide by precipitation from solutions containing aluminium salts
5/24	• Magnesium carbonates	7/36	• • • from organic aluminium salts
5/26	• Magnesium halides	7/38	• • Preparation of aluminium oxide by thermal reduction of aluminous minerals
5/28	• • Fluorides	7/40	• • • in the presence of aluminium sulfide
5/30	• • Chlorides	7/42	• • Preparation of aluminium oxide or hydroxide from metallic aluminium, e.g. by oxidation
5/32	• • • Preparation of anhydrous magnesium chloride by chlorinating magnesium compounds	7/44	• • Dehydration of aluminium hydroxide
5/34	• • • Dehydrating magnesium chloride containing water of crystallisation	7/46	• • Purification of aluminium oxide, aluminium hydroxide or aluminates [5]
5/36	• • Bromides	7/47	• • • of aluminates [5]
5/38	• Magnesium nitrates	7/48	• Aluminium halides
5/40	• Magnesium sulfates (double sulfates of magnesium with sodium or potassium C01D 5/12, with other alkali metals C01D 15/06, C01D 17/00) [3]	7/50	• • Fluorides
5/42	• Magnesium sulfites	7/52	• • • Double compounds containing both fluorine and other acid groups
		7/54	• • • Double compounds containing both aluminium and alkali metals or alkaline earth metals
7/00	Compounds of aluminium	7/56	• • Chlorides (containing fluorine C01F 7/52) [3]
7/02	• Aluminium oxide; Aluminium hydroxide; Aluminates	7/58	• • • Preparation of anhydrous aluminium chloride
7/04	• • Preparation of alkali metal aluminates; Aluminium oxide or hydroxide therefrom	7/60	• • • • from oxygen-containing aluminium compounds

C01F

- 7/62 • • • Purification
- 7/64 • • Bromides (containing fluorine C01F 7/52) [3]
- 7/66 • Aluminium nitrates (containing fluorine C01F 7/52) [3]
- 7/68 • Aluminium compounds containing sulfur (containing fluorine C01F 7/52) [3]
- 7/70 • • Sulfides
- 7/72 • • Sulfites
- 7/74 • • Sulfates
- 7/76 • • • Double salts, e.g. alums
- 11/00 Compounds of calcium, strontium, or barium**
(C01F 7/00 takes precedence) [3]
- 11/02 • Oxides or hydroxides (production of lime C04B 2/00)
- 11/04 • • by thermal decomposition
- 11/06 • • • of carbonates
- 11/08 • • by reduction of sulfates
- 11/10 • • from sulfides
- 11/12 • • from silicates
- 11/16 • • Purification
- 11/18 • Carbonates
- 11/20 • Halides
- 11/22 • • Fluorides
- 11/24 • • Chlorides
- 11/26 • • • from sulfides
- 11/28 • • • by chlorination of alkaline earth metal compounds
- 11/30 • • • Concentrating; Dehydrating; Preventing the absorption of moisture or caking
- 11/32 • • • Purification
- 11/34 • • Bromides
- 11/36 • Nitrates
- 11/38 • • Preparation with nitric acid or nitrogen oxides
- 11/40 • • Preparation by double decomposition with nitrates
- 11/42 • • Double salts (with magnesium C01F 5/38)
- 11/44 • • Concentrating; Crystallising; Dehydrating; Preventing the absorption of moisture or caking
- 11/46 • Sulfates (dehydration of gypsum C04B 11/02)
- 11/48 • Sulfites
- 13/00 Compounds of radium**
- 15/00 Compounds of thorium**
- 17/00 Compounds of the rare-earth metals, i.e. scandium, yttrium, lanthanum, or the group of the lanthanides**

C01G COMPOUNDS CONTAINING METALS NOT COVERED BY SUBCLASSES C01D OR C01F (metal hydrides C01B 6/00; salts of oxyacids of halogens C01B 11/00; peroxides, salts of peroxyacids C01B 15/00; thiosulfates, dithionites, polythionates C01B 17/64; compounds containing selenium or tellurium C01B 19/00; binary compounds of nitrogen with metals C01B 21/06; azides C01B 21/08; metal amides C01B 21/092; nitrites C01B 21/50; phosphides C01B 25/08; salts of oxyacids of phosphorus C01B 25/16; carbides C01B 31/30; compounds containing silicon C01B 33/00; compounds containing boron C01B 35/00; compounds having molecular sieve properties but not having base-exchange properties C01B 37/00; compounds having molecular sieve and base-exchange properties, e.g. crystalline zeolites, C01B 39/00; cyanides C01C 3/08; salts of cyanic acid C01C 3/14; salts of cyanamide C01C 3/16; thiocyanates C01C 3/20; fermentation or enzyme-using processes for the preparation of elements or inorganic compounds except carbon dioxide C12P 3/00; obtaining metal compounds from mixtures, e.g. ores, which are intermediate compounds in a metallurgical process for obtaining a free metal C21B, C22B; production of non-metallic elements or inorganic compounds by electrolysis or electrophoresis C25B)

Note(s)

1. Attention is drawn to Note (1) after class C01, which defines the last place priority rule applied in this class, i.e. in the range of subclasses C01B-C01G and within these subclasses.
2. Therapeutic activity of compounds is further classified in subclass A61P.

Subclass index

GENERAL METHODS OF PREPARATION.....	1/00
METALLIC COMPOUNDS, IN ALPHABETICAL ORDER OF THE SYMBOL FOR THE METAL	
Ag Silver.....	5/00
As Arsenic.....	28/00
Au Gold.....	7/00
Bi Bismuth.....	29/00
Cd Cadmium.....	11/00
Co Cobalt.....	51/00
Cr Chromium.....	37/00
Cu Copper.....	3/00
Fe Iron.....	49/00
Ga Gallium.....	15/00
Ge Germanium.....	17/00
Hf Hafnium.....	27/00
Hg Mercury.....	13/00
In Indium.....	15/00
Ir Iridium.....	55/00
Mn Manganese.....	45/00
Mo Molybdenum.....	39/00
Nb Niobium.....	33/00
Ni Nickel.....	53/00
Os Osmium.....	55/00
Pb Lead.....	21/00

Pd Palladium.....	55/00
Pt Platinum.....	55/00
Re Rhenium.....	47/00
Rh Rhodium.....	55/00
Ru Ruthenium.....	55/00
Sb Antimony.....	30/00
Sn Tin.....	19/00
Ta Tantalum.....	35/00
Ti Titanium.....	23/00
Tl Thallium.....	15/00
U Uranium.....	43/00
V Vanadium.....	31/00
W Tungsten.....	41/00
Zn Zinc.....	9/00
Zr Zirconium.....	25/00
COMPOUNDS OF TRANSURANIC ELEMENTS.....	56/00
COMPOUNDS OF METALS NOT COVERED BY THE PRECEDING GROUPS.....	99/00

1/00	Methods of preparing compounds of metals not covered by subclasses C01B, C01C, C01D, C01F, in general (electrolytic production of inorganic compounds C25B 1/00) [2]	19/00	Compounds of tin
1/02	• Oxides	19/02	• Oxides
1/04	• Carbonyls	19/04	• Halides
1/06	• Halides	19/06	• • Stannous chloride
1/08	• Nitrates	19/08	• • Stannic chloride
1/10	• Sulfates	21/00	Compounds of lead
1/12	• Sulfides	21/02	• Oxides
1/14	• Sulfites	21/04	• • Lead suboxide (Pb ₂ O)
3/00	Compounds of copper	21/06	• • Lead monoxide (PbO)
3/02	• Oxides; Hydroxides	21/08	• • Lead dioxide (PbO ₂)
3/04	• Halides	21/10	• • Red lead (Pb ₃ O ₄)
3/05	• • Chlorides [3]	21/12	• Hydroxides
3/06	• • Oxychlorides	21/14	• Carbonates
3/08	• Nitrates	21/16	• Halides
3/10	• Sulfates	21/18	• Nitrates
3/12	• Sulfides	21/20	• Sulfates
3/14	• Complexes with ammonia	21/21	• Sulfides [3]
5/00	Compounds of silver	21/22	• Plumbates; Plumbites
5/02	• Halides [3]	23/00	Compounds of titanium
7/00	Compounds of gold	23/02	• Halides of titanium
9/00	Compounds of zinc	23/04	• Oxides; Hydroxides [3]
9/02	• Oxides; Hydroxides [3]	23/047	• • Titanium dioxide [3]
9/03	• • Processes of production using dry methods, e.g. vapour phase processes [3]	23/053	• • • Producing by wet processes, e.g. hydrolysing titanium salts [3]
9/04	• Halides	23/07	• • • Producing by vapour phase processes, e.g. halide oxidation [3]
9/06	• Sulfates	23/08	• • • Drying; Calcining [3]
9/08	• Sulfides	25/00	Compounds of zirconium
11/00	Compounds of cadmium	25/02	• Oxides
11/02	• Sulfides [3]	25/04	• Halides
13/00	Compounds of mercury	25/06	• Sulfates
13/02	• Oxides	27/00	Compounds of hafnium
13/04	• Halides	27/02	• Oxides
15/00	Compounds of gallium, indium, or thallium	27/04	• Halides
17/00	Compounds of germanium	27/06	• Sulfates
17/02	• Germanium dioxide	28/00	Compounds of arsenic [3]
17/04	• Halides of germanium	28/02	• Arsenates; Arsenites [3]
		29/00	Compounds of bismuth
		30/00	Compounds of antimony [3]
		30/02	• Antimonates; Antimonites [3]

C01G

31/00 Compounds of vanadium

- 31/02 • Oxides [3]
- 31/04 • Halides [3]

33/00 Compounds of niobium

35/00 Compounds of tantalum

- 35/02 • Halides [3]

37/00 Compounds of chromium

- 37/02 • Oxides or hydrates thereof
- 37/027 • • Chromium dioxide [3]
- 37/033 • • Chromium trioxide; Chromic acid [3]
- 37/04 • Chromium halides
- 37/06 • • Chromylhalides
- 37/08 • Chromium sulfates
- 37/10 • • Chrome alum
- 37/14 • Chromates; Bichromates

39/00 Compounds of molybdenum

- 39/02 • Oxides; Hydroxides [3]
- 39/04 • Halides [3]
- 39/06 • Sulfides [3]

41/00 Compounds of tungsten

- 41/02 • Oxides; Hydroxides [3]
- 41/04 • Halides [3]

43/00 Compounds of uranium

- 43/01 • Oxides; Hydroxides [3]
- 43/025 • • Uranium dioxide [3]
- 43/04 • Halides of uranium
- 43/06 • • Fluorides
- 43/08 • • Chlorides
- 43/10 • • Bromides
- 43/12 • • Iodides

45/00 Compounds of manganese

- 45/02 • Oxides; Hydroxides
- 45/04 • Carbonyls
- 45/06 • Halides

- 45/08 • Nitrates
- 45/10 • Sulfates
- 45/12 • Manganates; Permanganates

47/00 Compounds of rhenium

49/00 Compounds of iron

- 49/02 • Oxides; Hydroxides
- 49/04 • • Ferrous oxide (FeO)
- 49/06 • • Ferric oxide (Fe₂O₃)
- 49/08 • • Ferroso-ferric oxide (Fe₃O₄)
- 49/10 • Halides
- 49/12 • Sulfides
- 49/14 • Sulfates
- 49/16 • Carbonyls

51/00 Compounds of cobalt

- 51/02 • Carbonyls
- 51/04 • Oxides; Hydroxides
- 51/06 • Carbonates
- 51/08 • Halides
- 51/10 • Sulfates
- 51/12 • Complexes with ammonia

53/00 Compounds of nickel

- 53/02 • Carbonyls
- 53/04 • Oxides; Hydroxides
- 53/06 • Carbonates
- 53/08 • Halides
- 53/09 • • Chlorides [3]
- 53/10 • Sulfates
- 53/11 • Sulfides [3]
- 53/12 • Complexes with ammonia

55/00 Compounds of ruthenium, rhodium, palladium, osmium, iridium, or platinum

56/00 Compounds of transuranic elements

99/00 Subject matter not provided for in other groups of this subclass [2010.01]